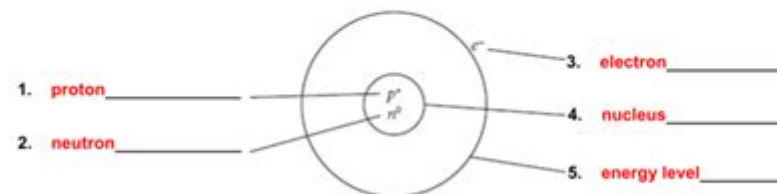


Name KEY Date \_\_\_\_\_ Class \_\_\_\_\_**Study Guide****CHAPTER 6****Section 1: Atoms, Elements, and Compounds****In your textbook, read about the structure of atoms.**

Label the diagram of an atom. Use these choices:

electron      energy level      neutron      nucleus      proton

**In your textbook, read about elements, compounds, and chemical bonds.**

If the statement is true, write true. If the statement is false, replace the italicized term or phrase to make it true.

6. On the periodic table, each element has a unique name and
- formula*
- .

Symbol

7. The periodic table is organized into horizontal rows, called periods, and vertical columns, called
- elements*
- .

group

- 8.
- Water*
- is composed of hydrogen and oxygen.

true

9. Atoms of the same element that have different numbers of neutrons are called
- isotopes*
- .

true

10. The
- period*
- of an element is the amount of time it takes for half of a radioactive isotope to decay.

Half-life

11. A
- combination*
- is a substance formed when two or more different elements combine.

compound

12. The two main types of chemical bonds are
- covalent bonds*
- and
- van der Waals forces*
- .

Covalent bond and Ionic bond

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